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Education

Ph. D. (March, 1984), M. Eng. (March, 1981), B. Eng. (March, 1979): Osaka University

Academic Carrier

1984 (April)-1989 (August): Assistant Professor, The Institute of Scientific and Industrial Research, Osaka University
1988 (May)-1989 (August): Post-doctoral fellow, Department of Chemistry, University of Illinois, Urbana-Champaign (supervisor: Prof. Scott E. Denmark)
1989 (September): Assistant Professor, Department of Applied Chemistry, Faculty of Engineering, Osaka University
1992 (September): Associate Professor, Department of Applied Chemistry, Faculty of Engineering, Osaka University
2003 (April)-: Professor, Department of Applied Chemistry, Faculty of Engineering, Osaka University
2007 (April)-: Director of Research Center for Environmental Preservation
2011-: A member of Science Council of Japan

Awards and Honors

Young Chemist Award of the Chemical Society of Japan (CSJ) (1991), The fourth Green and Sustainable Chemistry Award: Minister of Education, Sports, Culture, Science and Technology Prize (2005)

Total Publications

(SCI: 196), Citation (SCI): 7,447 (2011, August), h-index: 46

Research Interests

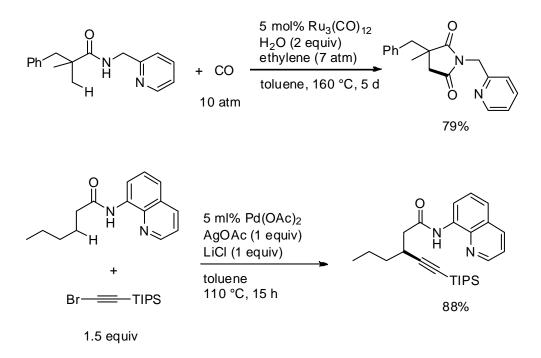
Development of new catalytic reactions

Transformation of Unactivated C(sp³)-H Bonds

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The direct utilization of C-H bonds, which are ubiquitous in organic molecules, is a straightforward method in organic synthesis which avoids the need for the prefunctionalization of starting materials. The utilization of C-H bonds involving the activation of $C(sp^2)$ -H bonds by transition metal complexes is now a commonly used method in organic synthesis. In fact, a wide variety of catalytic transformations of C-H bonds in arenes, heteroarenes, and alkenes have already been reported. In contrast, functionalization involving the activation of $C(sp^3)$ -H bonds continues to be a challenge in organic synthesis.¹ I wish to discuss two types of catalytic transformations of unactivated $C(sp^3)$ -H bonds; carbonylation² and ethynylation.³ In both cases, a choice of bidentate directing groups is important for the reaction to proceed selectively.



Jazzar, R.; Hitce, J.; Renaudat, A.; Sofack-Kreutzer, J.; Baudoin, O. Chem. Eur. J. 2010, 16, 2654-2672. Li, H.; Li, B.-J.; Shi, Z.-J. Catal. Sci. Technol. 2011, 1, 191-206.

⁽²⁾ Hasegawa, N.; Charra, V.; Inoue, S.; Fukumoto, Y.; Chatani, N. J. Am. Chem. Soc. 2011, 133, 8070-8073.

⁽³⁾ Ano, Y.; Tobisu, M.; Chatani, N. J. Am. Chem. Soc. 2011, 133, 12984-12986.